

## Solutions Manual Chemical Kinetics

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### Solutions Manual Chemical Kinetics

Topics and Subtopics in NCERT Solutions for Class 12 Chemistry Chapter 4 Chemical Kinetics: 4.1.For the reaction R $\rightarrow$ P, the concentration of a reactant changes from 0.03 M to 0.02 M in 25 minutes. Calculate the average rate of reaction using units of time both in minutes and seconds. 4.2.In a reaction, 2A  $\rightarrow$  Products, the concentration of A decreases from 0.5 mol L<sup>-1</sup> to 0.4 molL<sup>-1</sup> in 10 minutes.

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### Solution Manual & Test Bank: November 2016

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