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Flashcards | Quizlet Place the ions in their proper order: cation and then anion. Name the cation. Metals that form only one cation.

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Section 10.1 Measuring Matter Section
10.2 Mass and the Mole Section 10.3

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Section 10.5 Formulas of Hydrates Exit
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CHAPTER 10 REVIEW States of Matter

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SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting

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Place the ions in their proper order:
cation and then anion. Name the cation.
Metals that form only one cation. As
noted in Section 3.3, these metals are
usually in groups 1-3, 12, and 13. The
name of the cation of a metal that forms

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only one cation is the same as the name of the metal (with the word ion added if the cation is by itself). For example, Na^+ is the sodium ion, Ca^{2+} is the ...

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Section 10.3 - The ...

Section 3 (page 3) 1. He showed that maggots come from the eggs. laid by flies on meat, not from the meat itself. 2. He showed that a sealed flask of boiled broth. developed no tiny organisms. 3. He showed that broth became contaminated. only when it was exposed to the air. 4. Living things come

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spontaneously from nonliving matter. 5.

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Figure $\{\{3\}$: A Diels Alder reaction and its reverse. Typically, Diels Alder reactions occur at low or moderate temperatures (between 25 °C and 100 °C is a common range). In comparison, retro Diels Alder reactions require more elevated temperatures, often above 200 °C.

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